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Structural, magnetic, magnetocaloric and specific heat investigations on Mn doped PrCrO_3 orthochromites

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Abstract

We have synthesized $\text{PrCr}_{0.85}\text{Mn}_{0.15}\text{O}_3$ (PCMO) chromite and investigated the influence of manganese (Mn) doping at Cr-sites on the structural, magnetic, magnetocaloric and thermal properties of PrCrO_3 compound. No structural transition was observed with Mn substitution and the doped compound crystallized into distorted orthorhombic structure with $Pnma$ space group which was confirmed by Rietveld refinement of x-ray powder diffraction patterns. Neel temperature, noticed at 168 K from the temperature variation of magnetization, smaller than that reported for PrCrO_3 , indicated the influence of Mn^{3+} substitution in decreasing the antiferromagnetic ordering. Magnetization was almost eight times higher than that reported for undoped sample. Magnetocaloric effect measured via the magnetic entropy change and relative cooling power demonstrated significant values in the temperature range 10–20 K. The values of magnetic entropy change are much superior to that reported for other polycrystalline orthochromites and even at smaller applied field strength. The material exhibited second order magnetic phase transition. The Debye temperature and the density of states at Fermi level were also calculated. The overall results make $\text{PrCr}_{0.85}\text{Mn}_{0.15}\text{O}_3$ chromite a potential candidate to replace the existing materials for low temperature magnetic refrigeration.

Keywords: orthochromites, magnetic refrigeration, specific heat

(Some figures may appear in colour only in the online journal)

1. Introduction

Perovskite orthochromites are very promising materials for thermomagnetic switches, spintronic and thermally assisted magnetic random access memory (TAMRAM) devices [1–4]. Their general chemical formula is RCrO_3 (R = rare earth or yttrium ion) having distorted orthorhombic structure ($Pnma$ space group) and they exhibit canted antiferromagnetism in the G-type configuration below the Neel temperature (T_N) in the range 112–282 K which depends upon the ionic size of the

tri-valent rare-earth ion present [5–10]. They also show the temperature induced magnetization reversal (i.e. the magnetization changes its polarity from positive to negative at one particular temperature known as compensation temperature) and spin reorientation transition at low temperatures (below 50 K) [3, 4, 8, 11]. Below the Neel temperature, the spins of nearest Cr^{3+} ions do not possess the perfect antiparallel configuration, therefore, net magnetization is not zero and along with antiferromagnetic behaviour one observes weak ferromagnetic component also. This is known as canted antiferromagnetic

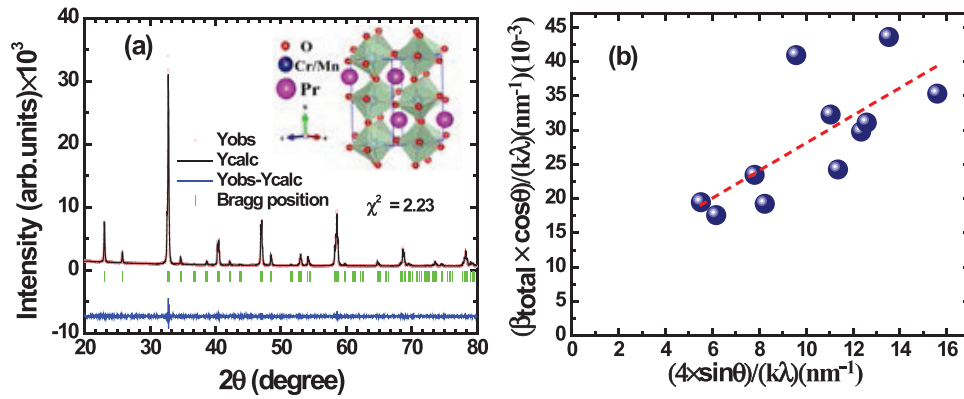


Figure 1. (a) The powder XRD profiles of PCMO showing the observed (red circle), calculated (black line), and difference intensities (blue line) along with the Bragg positions (green bars), inset shows the crystal structure of PCMO, (b) linear fit using the Williamson–Hall relation given by equation (1).

(CAFM) behaviour. Further, the Cr^{3+} ions produce the internal field at the rare-earth site [3, 6, 11]. Therefore, depending upon whether a magnetic ion, or non-magnetic ion is present at the rare-earth site, the internal field will (or will not) rotate the magnetic moment of the rare-earth ion in the opposite direction to that of Cr^{3+} ions. This way, the difference of the two magnetic moments will decide whether the particular orthochromite will demonstrate the magnetization reversal under low value of applied magnetic field. Whenever the applied magnetic field becomes greater than the internal field produced by chromium ions, the magnetization flipping along the field direction takes place resulting in the positive magnetization over the measurement range of temperature [4]. So far, many orthochromites like YCrO_3 [12–14], LaCrO_3 [15], CeCrO_3 [3, 16], PrCrO_3 [17–19], NdCrO_3 [20, 21], SmCrO_3 [22–24], GdCrO_3 [25, 26], DyCrO_3 [27, 28], YbCrO_3 [4, 29] and HoCrO_3 [30, 31] have been investigated for their complex magnetic properties and interesting results have been reported on them. As far as the PrCrO_3 (PCO) orthochromite is concerned, only a very few studies exist on it [17–19]. For example, Gordon *et al* measured its magnetic properties from liquid helium temperature to Neel temperature and obtained the latter ~ 237 K [17]. Similar studies have been reported by Venugopal Rao *et al* on polycrystalline powder prepared by citric acid route [19]. Liu *et al* have studied the effect of Ca^{2+} ion doping on the electrical properties of PrCrO_3 and proved the applicability of the compound for a superior solid oxide fuel cell (SOFC) [32]. Prasad *et al* reported the relaxor-like giant permittivity in PrCrO_3 and attributed it to the grain boundary effects [18]. From this, we conclude that only a few attempts have been made to tailor the structural and magnetic properties of PrCrO_3 chromite through site engineering which is commonly known as substitution or doping.

Recently, another interesting phenomenon called magnetocaloric effect (MCE) has been reported in chromites. Magnetic refrigeration (MR) measured through the indirect measurement of MCE in the cryogenic temperature range has been carried out systematically and carefully. In fact, giant MCE has been observed near the magnetic phase transitions in rare-earth chromites and multiferroic manganites [26, 31, 33, 34].

In this paper, we have investigated the influence of Mn ion doping on the structural magnetic, magnetocaloric and

specific heat properties of PrCrO_3 compound. To the best of our knowledge, this is for the first time that such studies are being reported on Mn doped PrCrO_3 orthochromites.

2. Experimental details

Polycrystalline samples of $\text{PrCr}_{0.85}\text{Mn}_{0.15}\text{O}_3$ (PCMO) were prepared by the solid state reaction method in ambient conditions by taking high purity Pr_6O_{11} , MnO_2 and Cr_2O_3 as starting materials (all chemicals from Sigma Aldrich). The mixture was ground thoroughly and first calcined for 24 h at 600°C followed by crushing and second calcination at 900°C for 12 h. After second calcination the powder was pulverized and pelletized in the form of uniform and compact pellets, which were sintered at 1350°C for 24 h in air. The phase identification was carried out by x-ray diffractometer (PANalytical-Empyrean) using $\text{Cu K}\alpha$ radiation with wavelength 1.540 \AA at a scanning rate of 0.02° from $20^\circ \leq \theta \leq 80^\circ$. The magnetic measurements were observed by vibrating sample magnetometer (VSM) between 4 K–300 K. The isotherms $M(H)$ measurements were measured between 0 and 90 kOe from 2 K to 102 K.

3. Result and discussions

3.1. Structural analysis

Figure 1(a) shows the x-ray diffraction patterns of sintered PCMO sample powder along with the Rietveld refinement using *Fullprof Suite* software [35]. The material is G-type ($a > b/\sqrt{2} > c$) orthorhombic perovskite ($Pnma$ space group) single phase in nature with lattice parameters $a = 5.491 \pm 0.002 \text{ \AA}$, $b = 7.7167 \pm 0.004 \text{ \AA}$. In place of this symbol please use the symbol like after 1.540 in the 11th line in the Experimental details. Please use that symbol everywhere in paper in place this symbol and $c = 5.4525 \pm 0.003 \text{ \AA}$. The lattice volume of PCMO unit cell is larger than that reported for pristine PrCrO_3 sample [36] which can be attributed to the larger size of Mn^{+3} (0.645 \AA) ion when compared with Cr^{+3} (0.615 \AA) ion. The results of the refinement are summarized in table 1.

Table 1. Structural parameters and atomic positions for PCMO powder at room temperature.

$R_p = 11.9\%$, $R_{wp} = 9.45\%$, $R_{exp} = 6.33\%$, density = 7.294 g cm^{-3} , $S = (7.03 \pm 0.4) \times 10^{-3}$, B–O1 = 1.976 (\AA) , Pr–O1 = 2.3475 (\AA) , B–O1–B = 154.806° , and B–O2–B = 153.814°

Atom	x	y	z	Occ
Pr: $4c(x, 0.25, z)$	0.03664	0.25	−0.00536	1.0128
Cr: $4b(0, 0, 0.5)$	0	0	0.5	0.8461
Mn: $4b(0, 0, 0.5)$	0	0	0.5	0.1516
O1: $4c(x, 0.25, z)$	0.48616	0.25	0.07783	0.9984
O2: $8d(x, y, z)$	0.29633	0.04511	−0.27708	2.8443

The spin-lattice interaction dictates the spin configuration and the tilting of BO_6 octahedra (inset figure 1(a) made by VESTA software [37]) for perovskite cell, where B is Cr/Mn ion. In BO_6 octahedron, $3d$ degenerate orbitals split into two states due to crystal field splitting: one with the lower energy state and the other is higher energy state named as t_{2g} and e_g , respectively [38]. Further hyperfine splitting of e_g orbitals known as Jahn–Teller splitting is possible if these orbitals contain odd number of electrons. Such type of splitting leads to the spin-lattice coupling and ensuing tilting of BO_6 octahedra. The sizable mismatch between R–O and B–O bond lengths in the Goldschmidt tolerance factor $t \equiv (\text{R–O})/\sqrt{2} \times (\text{B–O})$ results in $t < 1$ and significant cooperative rotations of the BO_6 octahedra (the average ionic radii ($R_{b,avg}$) of the B-site ions is calculated by $\sqrt{0.15 \times r_{Mn}^2 + 0.85 \times r_{Cr}^2}$, i.e. $R_{b,avg} = 0.6196 \text{ \AA}$, considering the atomic number ratio and ionic radii of Cr^{3+} (0.615 \AA) and Mn^{3+} (0.645 \AA) ions). This leads to the bending of the B–O–B angle and hence the distortion in the structure is witnessed. The orthorhombic distortion of the unit cell from the ideal cubic structure, defined by the orthorhombic strain factor $S = 2(a - c)/(a + c)$ [39], was calculated and summarized in table 1 as well. It should also be mentioned here that with the addition of Mn at the Cr-site in PCMO, bond angles (in-plane B–O1–B and out-of-plane B–O2–B) increase which can be understood in terms of the bigger ionic size of Mn^{3+} than Cr^{3+} ion that provides a lesser space for the bond to tilt. All these results further indicate the incorporation of Mn ion at Cr-sites in PCMO compound.

Further, Williamson–Hall relation was utilized to calculate the crystallite size [40]:

$$\beta_{\text{total}} = \frac{k\lambda}{d_{\text{XRD}} \cos \theta} + 4\eta \tan \theta, \quad (1)$$

where β_{total} is the full width at half-maximum (FWHM) of the XRD peaks. It includes both instrumental and sample dependent broadening effects. k is the Debye–Scherrer constant (~ 0.94 for spherical crystallites), λ is the incident x-ray wavelength (1.540 \AA), θ is the diffraction angle, and η is the microstrain parameter. Figure 1(b) shows a plot of $\beta_{\text{total}} \cos \theta / k\lambda$ and $4\sin \theta / k\lambda$, the estimated values of the average crystallite size d_{XRD} and the microstrain parameter η are $125 \pm 14 \text{ nm}$ and $(2.02 \pm 0.68) \times 10^{-3}$, respectively. The order of microstrain value fairly matches with the orthorhombic distortion S calculated before.

3.2. Magnetic properties

Figure 2(a) demonstrates the magnetization versus temperature variation under 200 Oe applied field in the field cooled (FC) mode. One can notice that magnetization is positive throughout the measurement range and increases with decreasing temperature. The maximum value of magnetization for the present compound is almost eight times higher when compared with that of the undoped PrCrO_3 material's magnetization even when it was measured under 1 T magnetic field [19]. This comparison attests the influence of Mn doping in enhancing the magnetic properties of chromites at low temperatures. In order to identify any magnetic transition, the derivative of magnetization with temperature i.e. dM/dT was plotted against temperature and the results are shown in figure 2(b) wherefrom two transitions at $T_{N1} \sim 197 \text{ K}$ (from paramagnetic to antiferromagnetic state and usually defined as Neel temperature) and $T_{N2} \sim 17 \text{ K}$ can be discerned. It is worth mentioning here that the Neel temperature of the undoped PCO sample is $\sim 237 \text{ K}$ [17], whereas in the present case it decreases to 197 K with 15% of Mn doping. This can be understood in terms of the appearance of double exchange interaction between Mn^{3+} and Cr^{3+} ions via O^{2-} leading to decrease in antiferromagnetic superexchange interaction. Thereby, T_{N1} can be expected to decrease with Mn-doping. Further, T_{N2} in figure 2(b) is assigned as a spin reorientation temperature due to ordering of Pr^{3+} – Pr^{3+} ions spins ($4f^2$; effective paramagnetic moment $\mu_{\text{eff}} = 3.58 \mu_B$). It is also argued that there is a spin structure change from the high temperature canted Γ_4 to collinear Γ_2 at T_{N2} . Figure 2(b) also depicts the inverse of susceptibility against temperature and it indicates that the high temperature paramagnetic region satisfies the Curie–Weiss behaviour ($\chi^{-1} = \frac{C}{T - \theta_{CW}}$). The calculated effective magnetic moment was $7.45 \mu_B$ (from the Curie constant C) along with Curie temperature (θ_{CW}) of -184 K , which is indicative for antiferromagnetic interactions. However, orthochromites like the present one possess distorted orthorhombic structure and the Cr–O–Cr bond angle θ deviates from 180° . Consequently, it results in the imperfect superexchange interaction and the ensuing CAFM (ferromagnetic + antiferromagnetic) behaviour below the Neel temperature. On comparing effective magnetic moment and Curie temperature values for the present PCMO compound with undoped PCO one [19] ($4.87 \mu_B$ and -158 K), we can clearly distinguish the contribution of manganese ions in magnetic behavior.

In order to confirm the coexistence of ferromagnetic and antiferromagnetic phases below the Neel temperature, the isothermal magnetization $M(H)$ loops at several characteristic temperatures 2, 50, 100, 150, 200 and 300 K were acquired in zero field mode (ZFC) and the graphs are shown in figure 2(c). The loops which are symmetric about the field axis exhibit hysteresis but without any saturation. The magnetization increases linearly for all the isothermal loops in the region of larger applied magnetic field. Such loops are also attributed to the coexistence of AFM (high field) and weak FM (low field) states. Several factors contribute to the observation of combined ferromagnetic and antiferromagnetic behaviors below the Néel temperature for the present compound.

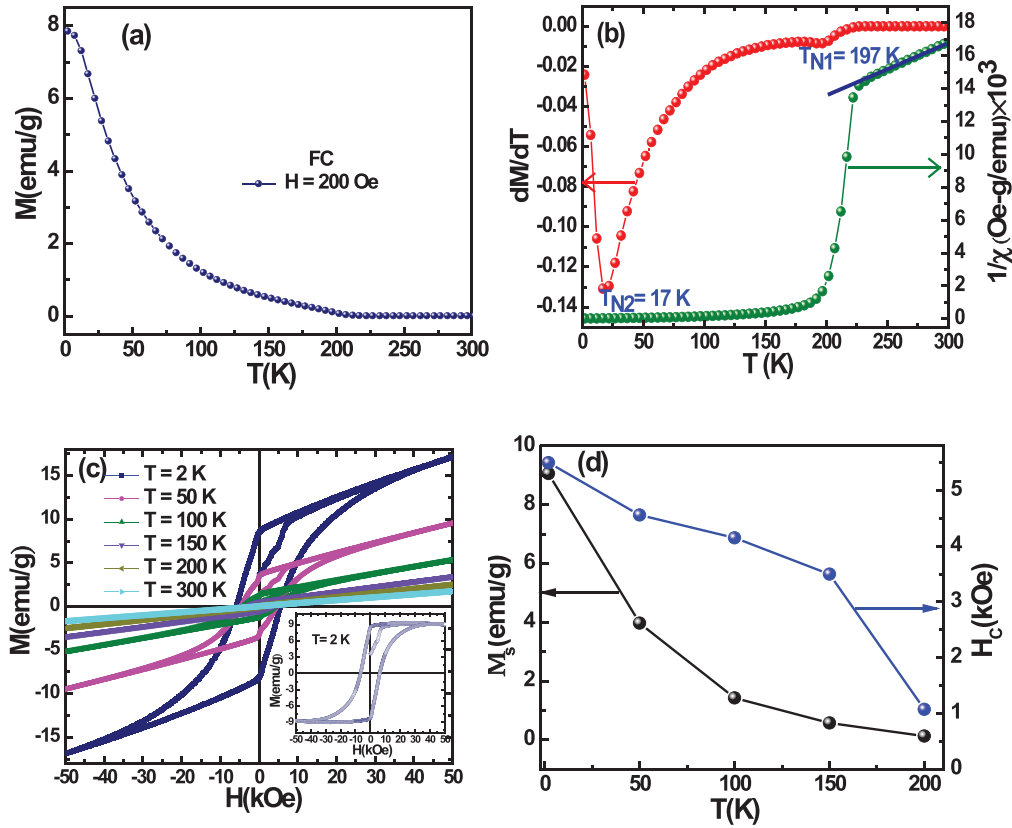


Figure 2. (a) $M(T)$ measured at $H = 200$ Oe in FC mode. (b) $\chi^{-1}(T)$ calculated at $H = 200$ Oe, along with dM/dT graph. (c) Isothermal $M(H)$ curves at temperatures 2, 50, 100, 150, 200 and 300 K (the inset is weak FM component versus magnetic field at 2 K), (d) the temperature dependence of the saturation magnetization and the coercivity H_c of the weak FM component.

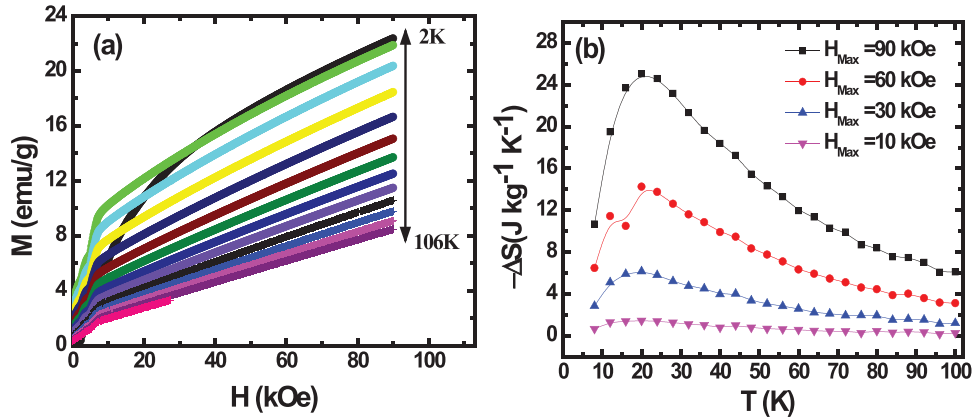


Figure 3. (a) Isothermal magnetization curves, (b) magnetic entropy change versus temperature at different applied magnetic field values.

For example, below T_{N1} , due to Dzialoshinski–Moriya type antisymmetric exchange interaction, Cr^{3+} ions are in the canted state resulting in the presence of ferromagnetic and antiferromagnetic states [41, 42]. Moreover, the interaction between Mn^{3+} ($t_{2g}^3 e_g^1$) and Cr^{3+} ($t_{2g}^3 e_g^0$) ions via O^{2-} would be ferromagnetic as Mn doping results in the decrease of Néel temperature [43–46]. As the temperature decreases further, f -orbitals in Pr^{3+} ions and p -orbitals of oxygen also start interacting and well below the Néel temperature, the Pr^{3+} – O^{2-} superexchange interaction also increases, resulting into the onset of Cr^{3+} – O^{2-} – Pr^{3+} interaction. This interaction

is ferromagnetic as here the empty e_g orbital of Cr^{3+} ions overlaps with one end of oxygen p -orbital and the other end of the same overlaps with a partially filled f -orbital of Pr^{3+} ion. This weak ferromagnetic superexchange coupling gives rise to opening up of hysteresis loop below 197 K (shown in figure 2(a)) with decreasing temperature and the remnant magnetization also increases [25]. The net magnetization is thus given by the relation $M(H) = \chi_{AF}H + M_s$, where $\chi_{AF}H$ is the antiferromagnetic contribution and M_s is the saturation magnetization of the weak ferromagnetism [47]. The value of M_s can be obtained by subtracting the antiferromagnetic

Table 2. Magnetic entropy change and relative cooling power at various applied magnetic field.

H_{Max} (kOe)	$-\Delta S_{\text{Max}}$ (J kg ⁻¹ K ⁻¹)	RCP (J kg ⁻¹)
10	1.45	34.669
30	6.16	153.076
50	11.44	422.136
60	14.25	526.53
90	25.04	708.231

contribution from the total magnetization. Inset of figure 2(c) shows the FM contribution for PCMO with the coercive field $H_c = 5.5$ kOe, and saturation magnetization of weak FM component is $M_s = 9.06$ emu g⁻¹ at $T = 2$ K. The values of the M_s and H_c decrease with increasing temperature up to the Neel temperature (figure 2(d)), and thereafter the hysteresis loops apparently disappears, indicating the phase transition to paramagnetic state.

3.3. Magnetocaloric measurements

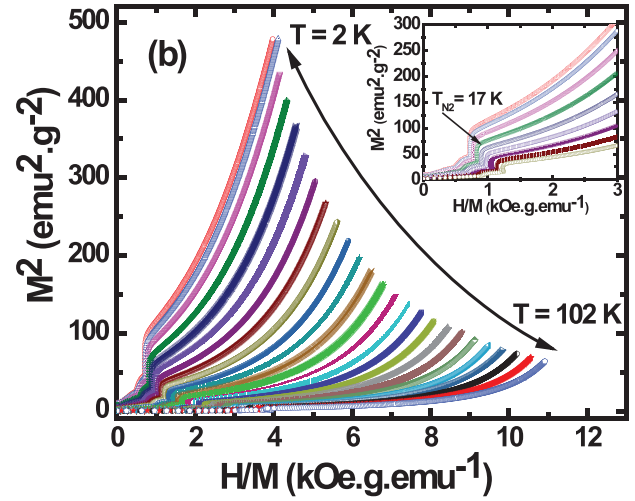
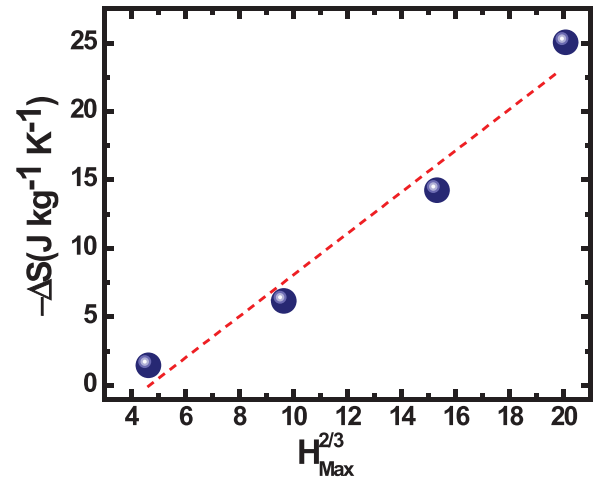
In order to further examine the performance of PCMO compound for MR, $M(H)$ data was acquired by varying the magnetic field from 0 to 90 kOe. The data were acquired between 2 to 106 K temperature range with an interval of $\Delta T = 4$ K. The results are plotted in figure 3(a) where the data has been shown from 2 to 106 K at a temperature difference of 8 K for the sake of clarity. The change of magnetic entropy (ΔS) caused by MCE is given by [48]:

$$\Delta S(T)_{\Delta H} = \int_{H_i}^{H_f} \left(\frac{\partial M(T, H)}{\partial T} \right) dH \quad (2)$$

Generally, for ΔS calculation from a series of $M-H$ curves, numerical integration of the above Equation by trapezoidal rule results [49]:

$$\Delta S(T_{\text{av}})_{\Delta H} = \frac{\delta H}{2 \delta T} \left(\delta M_1 + 2 \sum_{i=2}^{n-1} \delta M_i + \delta M_n \right), \quad (3)$$

where the average temperature $T_{\text{av}} = (T_j + T_{j+1})/2$ from the two magnetization isotherms measured at T_j and T_{j+1} in a magnetic field changing by $\Delta H = H_f - H_i$ at a constant step δH . While $\delta T = T_{j+1} - T_j$ is the temperature difference between the two isotherms, n is the number of points measured for each of the two isotherms with the magnetic field changing from $H_1 = H_i$ to $H_n = H_f$ at $\delta H = \Delta H/(n-1)$. Further $\delta M_i = [M(T_{j+1})_i - M(T_j)_i]$ is the difference in the magnetization at T_{j+1} and T_j for each magnetic-field step from 1 to n . It is worthwhile mentioning here that the sign of $-\Delta S$ provides information about the magnetic transition [50]. For a ferromagnetic (FM) transition, a positive value is obtained whereas an antiferromagnetic ordering displays a negative value due to an orientation disorder of the magnetic sublattices. Therefore, it is customary to plot values of $-\Delta S$ against T for different H values. The obtained results for the present PCMO compound are depicted in figure 3(b). It is clear that $-\Delta S$ is positive in the whole temperature measurement range with maximum value observed near T_{N2} . It suggests the peak due to FM

**Figure 4.** Arrott's plot for PCMO compound. Inset shows the zoomed view.**Figure 5.** Magnetic entropy variation (maximum) versus $H^{2/3}$.

ordering at T_{N2} . Further, $-\Delta S$ value is ~ 25.04 J kg⁻¹ K⁻¹ at 90 kOe. This value is two times higher than the one observed recently by our group in Mn doped SmCrO₃ compound [51]. The difference in the two values could be attributed to the higher magnetic moment associated with Pr³⁺ ions ($3.58 \mu_B$) when compared with that of Sm³⁺ ions ($0.84 \mu_B$). The magnetic entropy variation values for our PCMO compound are superior to (even at smaller applied magnetic field) the results reported on polycrystalline chromites and manganites in the temperature range 10–20 K [6, 31, 34]. For example, Yoshii [6] observed $-\Delta S$ value is ~ 4.6 J kg⁻¹ K⁻¹ in the 10–20 K temperature range under 5 T applied field whereas it is 11.44 J kg⁻¹ K⁻¹ for the compound investigated here (table 2). Yin *et al* [31] have reported $-\Delta S$ values 7.2 (at 20 K), 6.83 (at 20 K), 13.08 J kg⁻¹ K (at 5 K) for HoCrO₃, HoCr_{0.7}Fe_{0.3}O₃, and DyCr_{0.7}Fe_{0.3}O₃ samples, respectively under 7 T applied magnetic field. Again looking at the values shown in table 2 we notice better value even at 6 T magnetic field. Similarly, the present results are better than described by Shao *et al* on Ho doped DyMnO₃ manganites [34]. Further, our $-\Delta S$ values are slightly inferior to that obtained on GdCrO₃ single crystals, however, our sample is polycrystalline in nature: easy

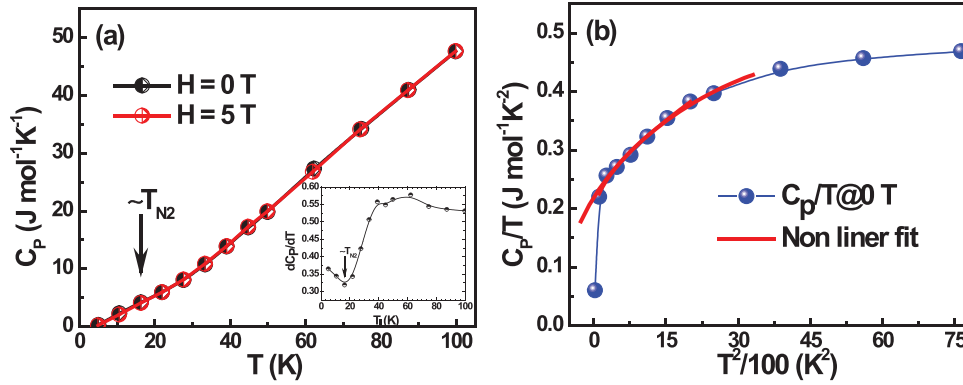


Figure 6. (a) Thermal evolution of heat capacity at zero and 5 T fields, (b) experimental C_p data is fitted according to the equation $C_p = \gamma T + \beta T^3 + \delta T^{3/2}$ at low temperature region in zero field condition.

to fabricate and cost effective when compared with single crystals.

In the present case, it is apparent that the rotation of Pr^{3+} ions with applied field and ordering between them brings positive value corresponding to a FM state in the T_{N2} region. It is also envisaged that the alignment of Mn^{3+} ions along the applied field direction contributes in the enhancement of $-\Delta S$ values. The interaction among the 3d electrons of Cr and Mn ions with 6f electrons of Pr ion also leads to enhancement of magnetic entropy change values. We further calculated the relative cooling power (RCP) or refrigerant capacity by the following expression:

$$\text{RCP} = |\Delta S^{\max}| \times \Delta T_{\text{FWHM}}, \quad (4)$$

where ΔT_{FWHM} is the full width at half maximum of the temperature dependent magnetic entropy change. RCP value is $\sim 371 \text{ J kg}^{-1}$ at the maximum field of 90 kOe. The RCP values at different fields (table 2) are quite high making the present compound one of the promising candidates for replacement of costlier, environment degrading, gas refrigerant materials in the liquid helium range.

Further the prediction about the order parameter of the magnetic transition could be made with the help of Arrott plot [52] between H/M versus M^2 . The first order phase transition yields negative slope whereas it is positive for second order. In case of PCMO compound, the slope of Arrott plot was found to be positive below T_{N2} , indicating the transition is of second order. The positive slope was observed above T_{N2} signifying that the magnetic transition nature is of second order (figure 3(b) and its inset shows a zoomed view). Thus PCMO compound exhibits large magnetic entropy change and RCP values with second order phase transition near the spin reorientation temperature.

For the materials undergoing a second-order transition, $-\Delta S^{\max}$ was predicted to develop as $H^{2/3}$ but later it was found that besides a term in $H^{2/3}$, $-\Delta S^{\max}$ contains an extra term independent of H [53]. This was explained on the basis of Landau's theory of second order phase transitions. The obtained expression was [53]:

$$-\Delta S^{\max} = \alpha \left(\frac{H}{4b} \right)^{2/3} - \frac{\alpha^2}{18b} \Delta T \quad (5)$$

ΔT is the width of the distribution of transition temperatures around the mean point, α and b are positive quantities independent of T or H that enter in the standard Landau expansion. Figure 5 shows as $-\Delta S^{\max} \propto H^{2/3}$ signalling that an approach based on mean field theory only is sufficient to explain the magnetic behaviour for high fields for the present PCMO sample.

3.4. Heat capacity

Since specific heat capacity (C_p) measurement is sensitive to the phase transition, therefore, to get further insight into the transition nature at T_{N2} , we measured C_p versus temperature without and with the application of 5 T magnetic field. The results are shown in figure 6. Though there is no anomaly apparent near T_{N2} , on plotting the derivative of specific heat with temperature i.e. dC_p/dT , a clear dip is observed at T_{N2} (inset to figure 6) which is a signature of the second order phase transition in the present studied sample. The heat capacity or specific heat (C_p) of a magnetic material is given by the equation [54]:

$$C_p = \gamma T + \beta T^3 + \delta T^{3/2} \quad (6)$$

where, γT describes the terms related to electronic contribution, βT^3 term gives the contribution from the lattice and $\delta T^{3/2}$ gives the magnetic contribution to C_p . Further, in the specific heat measurements large hyperfine interaction between the electronic and nuclear spins of magnetic rare-earth ion (in this case Pr^{3+}) commonly leads to a nuclear Schottky anomaly with a maximum at $\approx 0.3 \text{ K}$ [55, 56]. Since, in the present study specific heat measurement ranges from 5 to 100 K, therefore, we excluded the hyperfine contribution to the specific heat. The fitting of equation (6) to the data is shown in figure 6(b) and the values of the parameters obtained from the fitting are: $\gamma = 0.186 \text{ J} \cdot \text{K}^{-2} \text{ mole}^{-1}$, $\beta = 2.28 \times 10^{-5} \text{ J} \cdot \text{K}^{-4} \text{ mole}^{-1}$ and $\delta = 0.0034 \text{ J mole}^{-1} \text{ K}^{-3/2}$. From this fit it is clear that magnetic contribution to the total heat capacity is very small as compared to the electronic contribution, therefore, magnetic contribution (which in fact is smaller as we proved above) is masked in figure 6(a). Further, the electronic contribution, γ is related to the density of states (DOS) at the Fermi level: $\gamma = \pi^2 k_B^2 N(E_F)/3$, where $N(E_F)$ is the DOS.

Using $\gamma = 0.186 \text{ J} \cdot \text{K}^{-2} \text{ mole}^{-1}$ yielded a $N(E_F) = 3.2 \times 10^{25} \text{ eV}^{-1} \text{ cm}^{-3}$ which is in agreement with the results reported on manganites [54, 57]. The lattice contribution, β , is related to the Debye temperature (θ_D) through the relation $\beta = 234Nk_B/\theta_D^3$, where N is the number of ions per mole. The calculated value of θ_D was 440 K which is comparable to the $\text{Nd}_{1-x}\text{La}_x\text{CrO}_3$ system [58]. The overall magnitude of θ_D is 380–600 K for ABO_3 type perovskites [57, 59]. Since, heat capacity data on undoped PrCrO_3 compound is not available in the literature, therefore, it was not possible for us to bring out the difference caused by Mn doping in thermal parameters.

Finally, we could observe unprecedented high value of MCE in PMCO sample near the T_{N2} temperature along with the second order phase transition.

4. Conclusions

The impact of Mn doping on the structural, magnetic, magnetocaloric and specific heat properties of PCMO chromites has been investigated. Even with 15% Mn doping, the crystal structure of the compound remained orthorhombic structure with $Pnma$ space group, however, the lattice volume was found to increase when compared to that of pristine (undoped) PrCrO_3 sample. Neel transition temperature decreased with doping of Mn and the magnetization was almost eight times higher than reported for undoped sample. MCE measured through the magnetic entropy change and RCP showed higher values than reported for other polycrystalline chromites and manganites. The Debye temperature and the DOS at Fermi level were also calculated. The material exhibited second order magnetic phase transition between 10–20 K.

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